

## Thermodynamics An Engineering Approach Answers

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### Answer: Using the Gibbs function data, determine the ...

Solid carbon at 25°C is burned with a stoichiometric amount of air which is at 1 atm pressure and 25°C. Determine the number of moles of CO<sub>2</sub> formed per kmol of carbon when only CO<sub>2</sub>, CO, O<sub>2</sub>, and N<sub>2</sub> are present in the products and the products are at 1 atm and 727°C

### Answer: Solid carbon at 25°C is burned with a | StudySoup

P1=96.0 [kPa] P2=98.0 [kPa] rho=1.0 [kg/m<sup>3</sup>] g=9.81 [m/s<sup>2</sup>] DELTAP=P2-P1 DELTAP=rho\*g\*h\*Convert(Pa,kPa) "Some Wrong Solutions with Common Mistakes:" DELTAP=rho\*W1\_h/1000 "not using g" DELTAP=g\*W2\_h/1000 "not using rho" P2=rho\*g\*W3\_h/1000 "ignoring P1" P1=rho\*g\*W4\_h/1000 "ignoring P2" 1-123 Considera 2.5-m-deep swimmingpool. The pressure difference between the top and bottom of the pool is (a) 2.5 kPa (b) 12.0 kPa (c) 19.6 kPa (d) 24.5 kPa (e) 250 kPa Answer (d) 24.5 kPa Solution Solved by EES Software ...

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### Thermodynamics An Engineering Approach

While coming downhill the potential energy (P.E) of the bicyclist will be converted into Kinetic Energy (K.E), so the speed of the bicyclist increases. Total energy at any instant is given as. As the bicyclist coming downhill, h will decrease, therefore, the potential energy will decrease. But as the total energy is conserved, decrease in potential energy will be compensated by increase in kinetic energy.

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